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ChemQuest 21 Name: _____ Date: _____

Hour: _____ Information: Charges of Some Transition Elements So far you have learned that you can predict the charge that an ion will have based on its location on the periodic table. However, the transition elements are not easy to predict. A few common

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Information : Charges of Some Transition Elements

Many transition elements can have more than one charge when they become an ion. Copper ions, for example, can be Cu^+ or Cu^{2+} . As another example, iron ions are sometimes Fe^{2+} and sometimes Fe^{3+} . Critical Thinking Questions. 2. Copper and iron are in the “d block” and so you need to calculate their charge by comparing what bonds to them.

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Chemquest 22 Covalent Bonding Answer Key

However, the transition elements are not easy to predict. A few common transition elements are listed below. You should memorize their charges. Silver: Ag +
Zinc: Zn 2+ Cadmium: Cd 2+ Critical Thinking Questions 1.

(Chem 1)

CQ_26_Transition_Metals_with_SP - ChemQuest 26 ...

ChemQuest 13 Name: _____ Date: _____
Hour: _____ Information: Energy of
Sublevels Each sublevel has a different amount of energy. For example, orbitals in the 3p sublevel have more energy than orbitals in the 2p sublevel. The

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following is a list of the sublevels from lowest to highest energy:

Information: Hund's Rule

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